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# L-GLUTAMINE / AMMONIA (Rapid)

**ASSAY PROCEDURE** 

K-GLNAM 06/18

(\*50 Manual Assays Per Kit) or (500 Microplate Assays Per Kit)

\*The number of tests per kit can be doubled if all volumes are halved



## **INTRODUCTION:**

L-Glutamine is an essential component of certain cell culture media. However, the incorporation of this amino acid into growth media presents two major problems; firstly, L-glutamine is labile and spontaneously breaks down to L-glutamate and free ammonium ions, and secondly, the released ammonium ions are very toxic to the cells. To overcome these issues, either pre-formulated growth media are used strictly within their recommended shelf-lives or L-glutamine is added just before use. In either case, monitoring of L-glutamine and ammonia is frequently performed both prior to, and during, culturing.

Individual test kits do exist for L-glutamine and ammonia, but these products can be very slow, owing to either the limiting amounts or nature of the enzymes employed. Megazyme thus developed this kit (K-GLNAM), based on advanced recombinant enzymes, that is both very rapid (~ 15 min) and also measures both analytes in a very simple format. As the kit is based on quantitative endpoint readings, a calibration curve is not required, saving more time and expense. Both manual (see page 5, "A") and microplate (see page 8, "B") formats are described. Other rapid cell culture test kits are also available from Megazyme for ammonia (K-AMIAR), L-asparagine /ammonia (K-ASNAM), D-glucose (K-GLUHK or K-GLUC) and L-lactic acid (K-LATE).

This kit is also suitable for the analysis of L-glutamine in a wide range of other samples, such as dietary supplements and vegetables.

## **PRINCIPLE:**

Determination of L-glutamine takes place in two simple and rapid steps; L-glutamine is first deaminated in the presence of glutaminase to L-glutamate and ammonium ions  $(NH_4^+)$  (1).

(glutaminase)
(1) L-Glutamine + 
$$H_2O$$
  $\longrightarrow$  L-glutamate +  $NH_4^+$ 

In the presence of reduced nicotinamide-adenine dinucleotide phosphate (NADPH) and glutamate dehydrogenase (GIDH), the ammonia formed in reaction (I) reacts with 2-oxoglutarate to form L-glutamate and NADP<sup>+</sup> (2).

(GIDH) (2) 
$$NH_4^+ + 2$$
-oxoglutarate +  $NADPH \longrightarrow L$ -glutamate +  $NADP^+ + H_2O$ 

The amount of NADP<sup>+</sup> formed is stoichiometric with the amount of glutamine and ammonia. It is NADPH consumption that is measured by the decrease in absorbance at 340 nm.

# SPECIFICITY, SENSITIVITY, LINEARITY AND PRECISION:

The assay is specific for L-glutamine and free ammonium ions. The D-isomer does not react and the glutaminase employed has no activity on the related amino acid L-asparagine.

The smallest differentiating absorbance for the assay is 0.005 absorbance units. This corresponds to 0.271 mg/L of sample solution at the maximum sample volume of 1.00 mL. The detection limit is 0.543 mg/L, which is derived from an absorbance difference of 0.010 with a sample volume of 1.0 mL.

The assay is linear over the range of I to 40 µg of L-glutamine per assay. In duplicate determinations using one sample solution, an absorbance difference of 0.005 to 0.010 may occur. With a sample volume of 1.00 mL, this corresponds to a L-glutamine concentration of approx. 0.271 to 0.543 mg/L of sample solution. If the sample is diluted during sample preparation, the result is multiplied by the dilution factor, F. If in sample preparation, the sample is weighed, e.g. 10 g/L, a difference of 0.02 to 0.05 g/100 g can be expected.

# **INTERFERENCE:**

If quantitative deamination of L-glutamine is achieved within the time specified in the assay (i.e. approx. 5 min), it can be generally concluded that no interference has occurred. Interference of the indicator reaction (2) can be checked by adding ammonia (approx. 4  $\mu g$  in 0.1 mL) to the cuvette on completion of the reaction. A significant decrease in the absorbance should be observed.

Interfering substances in the sample being analysed can be identified by including an internal standard. Quantitative recovery of this standard would be expected. Losses in sample handling and extraction are identified by performing recovery experiments, i.e. by adding L-glutamine to the sample in the initial extraction steps.

#### **SAFETY:**

The general safety measures that apply to all chemical substances should be adhered to.

For more information regarding the safe usage and handling of this product please refer to the associated SDS that is available from the Megazyme website.

## KITS:

Kits suitable for performing 50 assays of each in manual format (or 500 assays of each in microplate format) are available from Megazyme. The kits contain the full assay method plus:

Bottle I: Buffer (II mL, pH 4.9) plus sodium azide

(0.02% w/v) as a preservative. Stable for > 2 years at 4°C.

Bottle 2: (x2) Buffer (25.5 mL, pH 8.0) plus sodium azide

(0.02% w/v) as a preservative. Stable for > 2 years at  $4^{\circ}$ C.

Bottle 3: (x2) NADPH. Lyophilised powder.

Stable for > 5 years below -10°C.

**Bottle 4:** Glutaminase suspension (1.1 mL).

Stable for > 2 years at 4°C.

**Bottle 5:** Glutamate dehydrogenase (2.2 mL).

Stable for > 2 years at 4°C.

Bottle 6: Ammonia standard solution (5 mL, 0.04 mg/mL) in

0.02% (w/v) sodium azide. Stable for > 2 years at 4°C.

**Bottle 7:** L-Glutamine control powder (~ 2 g).

Stable for > 2 years at  $4^{\circ}$ C.

# PREPARATION OF REAGENT SOLUTIONS/SUSPENSIONS:

- Use the contents of bottles I and 2 as supplied. Stable for > 2 years at 4°C.
- 3. Dissolve the contents of one of bottle 3 in 12 mL of distilled water. Stable for > 1 year at 4°C or stable for > 2 years below -10°C (to avoid repetitive freeze/thaw cycles, divide into appropriately sized aliquots and store in polypropylene tubes). Do not dissolve the contents of the second bottle until required.
- **4 & 5.** Use the contents of bottles 4 and 5 as supplied. Before opening for the first time, shake the bottles to remove any enzyme that may have settled on the rubber stopper. Subsequently, store the bottles in an upright position. Stable for > 2 years at 4°C.
- 6. Use the contents of bottle 6 as supplied. Stable for > 2 years at 4°C.
- Accurately weigh approx. 0.30 g of L-glutamine into a 1 L
  volumetric flask, fill to the mark with distilled water and mix
  thoroughly.

Stable for  $\sim 2$  months below -10°C.

**NOTE:** The ammonia and L-glutamine standard solution is only assayed where there is some doubt about the accuracy of the spectrophotometer being used or where it is suspected that inhibition is being caused by substances in the sample. The concentration of ammonia and L-glutamine is determined directly from the extinction coefficient of NADPH (page 6).

# **EQUIPMENT (RECOMMENDED):**

- 1. Volumetric flasks (50 mL, 100 mL and 500 mL).
- 2. Disposable plastic cuvettes (I cm light path, 3.0 mL).
- 3. Micro-pipettors, e.g. Gilson Pipetman<sup>®</sup> (200  $\mu$ L and 1000  $\mu$ L).
- 4. Positive displacement pipettor, e.g. Eppendorf Multipette<sup>®</sup>
  - with 5 mL Combitip<sup>®</sup> [to dispense 0.2 mL aliquots of buffer (solution 1)].
  - with 25 mL Combitip<sup>®</sup> (to dispense 1.5 mL aliquots of distilled water and 0.5 mL aliquots of solution 3).
- 5. Analytical balance.
- 6. Spectrophotometer set at 340 nm.
- 7. Vortex mixer (e.g. IKA® Yellowline Test Tube Shaker TTS2).
- 8. Stop clock.
- 9. Whatman No. I (9 cm) filter papers.

# A. MANUAL ASSAY PROCEDURE:

Wavelength: 340 nm

Cuvette: I cm light path (glass or plastic)

**Temperature:** ~ 25°C **Final volume:** 2.34 mL

**Sample solution:** 0.4-7.0 µg of ammonia per cuvette

or I-40 µg of L-glutamine per cuvette

(in 0.10-1.00 mL sample volume)

Read against air (without a cuvette in the light path) or against water

Pipette into cuvettes	Ammonia		L-Glutamine	
	Blank	Sample	Blank	Sample
solution I (buffer) sample solution suspension 4 (Glutaminase)	- - -	- 0.10 mL -	0.20 mL - 0.02 mL	0.20 mL 0.10 mL 0.02 mL
Mix* and incubate for 5 min at room temperature. Then add:				
distilled water (at ~ 25°C) solution 2 (buffer) solution 3 (NADPH)		1.72 mL 0.30 mL 0.20 mL	1.60 mL 0.30 mL 0.20 mL	1.50 mL 0.30 mL 0.20 mL
Mix* and read the absorbances of the solutions (A <sub>1</sub> ) after approx. 4 min. Then start the reaction by addition of:				
suspension 5 (GIDH)	0.02 mL	0.02 mL	0.02 mL	0.02 mL

 $Mix^*$  and read the absorbances of the solutions  $(A_2)$  at the end of the reaction (approx. 5 min). If the reaction has not stopped after 5 min, continue to read the absorbances at I min intervals until the absorbances remain the same.

<sup>\*</sup> for example with a plastic spatula or by gentle inversion after sealing the cuvette with a cuvette cap or Parafilm<sup>®</sup>.

# **CALCULATION:**

Determine the absorbance differences  $(A_1-A_2)$  for both blanks and samples. Subtract the absorbance difference of the blank from the absorbance difference of the corresponding sample, thereby obtaining the change in absorbance  $(\Delta A)$  resulting from the analyte in question, as follows:

# Determination of ammonia:

 $\Delta A_{ammonia} = (A_1 - A_2)_{ammonia \ sample} - (A_1 - A_2)_{ammonia \ blank}$ 

# **Determination of L-glutamine:**

To calculate  $\Delta A_{L-glutamine}$ , it is first necessary to calculate  $\Delta A_{(L-glutamine + ammonia)}$ , in order to account for the free ammonium ions in the sample:

 $\Delta A$ (L-glutamine + ammonia) =  $(A_1-A_2)$ L-glutamine sample -  $(A_1-A_2)$ L-glutamine blank

#### Then:

 $\Delta A_{L-glutamine} = \Delta A_{(L-glutamine + ammonia)} - \Delta A_{ammonia}$ 

The values of  $\Delta A_{ammonia}$  and  $\Delta A_{L-glutamine}$  should as a rule be at least 0.100 absorbance units to achieve sufficiently accurate results.

The concentration of ammonia and L-glutamine can be calculated as follows:

$$c = \frac{V \times MW}{\varepsilon \times d \times v} \times \Delta A \qquad [g/L]$$

#### where:

V = final volume [mL]

MW = molecular weight of analyte [g/mol]

ε = extinction coefficient of NADPH at 340 nm

=  $6300 [l \times mol^{-1} \times cm^{-1}]$ 

d = light path [cm]

v = sample volume [mL]

# It follows for ammonia:

c = 
$$\frac{2.34 \times 17.03}{6300 \times 1.0 \times 0.10} \times \Delta A_{\text{ammonia}}$$
 [g/L]

= 
$$0.06325 \times \Delta A_{ammonia}$$
 [g/L]

# for L-glutamine:

$$c = \frac{2.34 \times 146.1}{6300 \times 1.0 \times 0.10} \times \Delta A_{L-glutamine}$$
 [g/L]

= 
$$0.5427 \times \Delta A_{L-glutamine}$$
 [g/L]

If the sample has been diluted during preparation, the result must be multiplied by the dilution factor, F.

When analysing solid and semi-solid samples which are weighed out for sample preparation, the content (g/100 g) is calculated from the amount weighed as follows:

# Content of ammonia

$$= \frac{c_{ammonia} [g/L \text{ sample solution}]}{\text{weight}_{sample} [g/L \text{ sample solution}]} \times 100 \qquad [g/100 \text{ g}]$$

# Content of L-glutamine

$$= \frac{c_{L-glutamine} [g/L \text{ sample solution}]}{\text{weight}_{sample} [g/L \text{ sample solution}]} \times 100 \qquad [g/100 \text{ g}]$$

**NOTE:** These calculations can be simplified by using the Megazyme *Mega-Calc*<sup>TM</sup>, downloadable from where the product appears on the Megazyme website (www.megazyme.com).

# **B. MICROPLATE ASSAY PROCEDURE:**

Wavelength: 340 nm

**Microplate:** 96 well (e.g. clear, flat-bottomed, polypropylene)

Temperature:  $\sim 25^{\circ}\text{C}$ Final volume:  $234~\mu\text{L}$ 

**Sample solution:** 0.04-0.70 µg of ammonia per well

or 0.1-4.0 µg of L-glutamine per well

(in 10 µL sample volume)

Pipette into cuvettes	Ammonia		L-Glutamine	
	Blank	Sample	Blank	Sample
solution I (buffer) sample solution suspension 4 (Glutaminase)		- 10 μL -	20 μL - 2 μL*	20 μL 10 μL 2 μL*
Mix** and incubate for 5 min at room temperature. Then add:				
distilled water (at ~ 25°C) solution 2 (buffer) solution 3 (NADPH)	182 μL 30 μL 20 μL	172 μL 30 μL 20 μL	160 μL 30 μL 20 μL	150 μL 30 μL 20 μL
Mix* and read the absorbances of the solutions $(A_I)$ after approx. 4 min. Then start the reaction by addition of:				
suspension 5 (GIDH)	2 μL*	2 μL*	2 µL*	2 μL*

 $Mix^{**}$  and read the absorbances of the solutions (A<sub>2</sub>) at the end of the reaction (approx. 5 min). If the reaction has not stopped after 5 min, continue to read the absorbances at I min intervals until the absorbances remain the same.

# **CALCULATION:**

Perform the calculations as described on pages 6 and 7, after appropriate path-length adjustment to 10 mm. This can either be performed automatically by the plate reader (certain models only), or after manual determination of the true path-length (i.e. by simply performing a "manual" format assay of the control solution, and comparing the absorbance change to that of a reaction performed according to the "microplate" format). Alternatively, a standard curve can be used.

<sup>\*</sup> if preferred, dilute sufficient enzyme for the set of assays 10-fold with distilled water and add 20  $\mu$ L. Reduce the amount of water appropriately (i.e. by 18  $\mu$ L) to maintain the same final volume.

<sup>\*\*</sup> for example using microplate shaker, shake function on a microplate reader or repeated aspiration (e.g. using pipettor).

## **SAMPLE PREPARATION:**

# I. Sample dilution (Manual Format analysis).

The amount of L-glutamine present in the cuvette (i.e. in the 0.10 mL of sample being analysed) should range between 1 and 40  $\mu g$ . The sample solution must therefore be diluted sufficiently to yield a concentration between 0.01 and 0.40 g/L.

## **Dilution Table**

Estimated concentration of L-glutamine (g/L)	Dilution with water	Dilution factor (F)
< 0.4	No dilution required	I
0.4-4.0	l + 9	10
4-40	l + 99	100
> 400	l + 999	1000

If the value of  $\Delta A_{ammonia}$  or  $\Delta A_{L\text{-glutamine}}$  is too low (e.g. < 0.100), weigh out more sample or dilute less strongly. Alternatively, the sample volume to be pipetted into the cuvette can be increased up to 1.00 mL, making sure that the sum of the sample and distilled water components in the cuvette is either 1.60 mL (for L-glutamine) or 1.82 mL (for ammonia) and using the new sample volume in the equation.

# 2. Sample clarification.

Carrez reagents cannot be used for deproteinisation as their use results in significantly reduced recoveries. Perchloric can be used as an alternative (see specific example).

## 3. General considerations.

NB - these considerations are for the analysis of complex samples, such as foodstuffs and beverages. They are not relevant in the analysis of cell culture media/supernatants; for this application see "Sample preparation example (a)".

- (a) Liquid samples: clear, slightly coloured and approximately neutral, liquid samples can be used directly in the assay.
- **(b) Carbon dioxide:** samples containing significant quantities of carbon dioxide, such as beer, should be degassed by increasing the pH to approx. 7.4 with 2 M NaOH and gentle stirring, or by stirring with a glass rod.
- (c) Coloured samples: an additional sample blank, i.e. sample with no GIDH, may be necessary in the case of coloured samples.

- (d) Strongly coloured samples: if used undiluted, strongly coloured samples should be treated by the addition of 0.2 g of polyvinylpolypyrrolidone (PVPP)/10 mL of sample. Shake the tube vigorously for 5 min and then filter through Whatman No. I filter paper.
- **(e) Solid samples:** homogenise or crush solid samples in distilled water and filter if necessary.
- (f) Samples containing fat: extract such samples with hot water at a temperature above the melting point of the fat, e.g. in a 100 mL volumetric flask at 60°C. Adjust to room temperature and fill the volumetric flask to the mark with water. Store on ice or in a refrigerator for 15-30 min and then filter. Discard the first few mL of filtrate and use the clear supernatant (which may be slightly opalescent) for assay.
- (g) Samples containing protein: deproteinise samples containing protein by adding an equal volume of ice-cold I M perchloric acid with mixing. Centrifuge at 1,500 g for 10 min and neutralise the supernatant with I M KOH.

# **SAMPLE PREPARATION EXAMPLES:**

# (a) Determination of ammonia and L-glutamine in cell culture media/supernatants.

In general, the concentration of ammonia and L-glutamine in cell culture media/supernatants can be determined without any sample treatment (except clarification by centrifugation/filtering or dilution according to the dilution table, if necessary). Typically, no clarification or dilution is required, and a sample volume of 0.1 mL is satisfactory.

# (b) Determination of ammonia and L-glutamine in powdered dietary supplements.

In general, the concentration of ammonia and L-glutamine in dietary supplements, such as pharmaceutical grade L-glutamine, can be determined as follows: accurately weigh approximately 5 g of representative material into a 100 mL volumetric flask. After addition of approx. 60 mL of distilled water, stir the contents until fully dissolved or suspended, and fill up to the mark with distilled water. Mix and, if necessary, filter through Whatman No. I filter paper. Use the clear filtrate, with dilution according to the dilution table if necessary, for the assay. Typically, for pharmaceutical grade L-glutamine, a further dilution of 1:100 and sample volume of 0.1 mL are satisfactory.

# (c) Determination of ammonia and L-glutamine in sports nutrition and bakery products (e.g. snack bars).

Homogenise approx. 10 g of material and accurately weigh approx. 2 g into a 100 mL volumetric flask. Add 60 mL of distilled water, and incubate at 60°C for 5 min, or until fully suspended. Allow to equilibrate to room temperature and fill to the mark with distilled water (ensuring any fat containing layer is "above" the mark, and the aqueous layer is "at" the mark). Filter, discarding the first 3-5 mL and use the filtrate for the assay. Typically, no further dilution is required and a sample volume of 0.1 mL is satisfactory.

# (d) Determination of ammonia and L-glutamine in fruit and vegetable products/preparations (e.g. potato juice).

Accurately weigh approx. 10 g of representative material into a 100 mL Duran<sup>®</sup> bottle, add 20 mL of perchloric acid and homogenise for 2 min using an Ultra-turrax<sup>®</sup> or Polytron<sup>®</sup> homogeniser (or equivalent). Quantitatively transfer to a 40 mL glass beaker and adjust the pH to approx. 8.0 using 2 M KOH. Quantitatively transfer to a 100 mL volumetric flask and adjust to the mark with distilled water (ensuring the fat containing layer is "above" the mark, and the aqueous layer is "at" the mark). Store on ice for 20 min to precipitate potassium perchlorate and allow separation of the fat. Filter, discarding the first 3-5 mL and use the clear filtrate for the assay. *Typically, no further dilution is required and a sample volume of 0.2 mL is satisfactory.* 

# **REFERENCE:**

Lund, P. (1990). L-Glutamine and L-Glutamate. "Methods of Enzymatic Analysis" (Bergmeyer, H. U., ed.), 3rd ed., Vol. VIII, pp. 357-363, VCH Publishers (UK) Ltd., Cambridge, UK.

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